## MALEIC ANHYDRIDE

Common Synonyms cis-Butenedioic anhydride 2,5-Furanedione Toxilic anhydride		crystals or tablets	Molten; or solid Colorless Choking odor crystals or tablets Sinks and mixes slowly with water.					
Wear gogg (including g Call fire de	les, self-conta loves). partment.	id contact with solid a ained breathing appar ollution control agenc	atus, and rubber overcloth	ning				
Fire	Combustible. Dust cloud may be ignited by spark or flame. Wear googles, self-contained breathing apparatus, and rubber overclothing (including gloves). Extinguish with dry chemical, alcohol foam, or carbon dioxide. Water may be ineffective on fire.							
Exposure	CALL FOR MEDICAL AID. LIQUID OR SOLID Will burn skin and eyes. Harmful if swallowed. Remove contaminated clothing and shoes. Flush affected areas with plenty of water. IF IN EYES, hold eyelids open and flush with plenty of water. IF SWALLOWED and victim is CONSCIOUS, have victim drink water or milk.							
Water Pollution	May be dan Notify local	Dangerous to aquatic life in high concentrations. May be dangerous if it enters water intakes. Notify local health and wildlife officials. Notify operators of nearby water intakes.						
1. CORRECTIVE Dilute and Stop disch Collection	disperse		2.1 CG Compatibili 2.2 Formula: OCOO 2.3 IMO/UN Designa 2.4 DOT ID No.: 221 2.5 CAS Registry N 2.6 NAERG Guide N	H=CHCO ation: 9.0/2215 5 o.: 108-31-6				
face shield 3.2 Symptoms Fol causes irri may occur. 3.3 Treatment of E water for a treat as ch 3.4 TLV-TWA: 0.23 3.5 TLV-STEL: Noi 3.5 TLV-STEL: Noi 3.6 TLV-Ceiling: N 3.7 Toxicity by Ing 3.8 Toxicity by Ing 3.9 Chronic Toxici 3.10 Vapor (Gas) Ir high coros 11 Liquid or Soli	r ubber glove lowing Expo attion and red exposure: INIt least 15 min isted. o t listed. ot listed. ot listed. ot listed. ot listed. for any constant currer to the state of characterized of the characterized of the form of	s and boots; coverall sure: Inhalation caus tess. Vapors cause : tALATION: give oxy; ; for eyes, call a phy armal burn. a 2; LDso = 0.5 to 5 g ently not available. teristics: Vapors ca assant. The effect is tics: causes smartir condary burns on lon ng/m <sup>3</sup> L	es coughing, sneezing, thi severe eye irritation; phot gen. EYE OR SKIN CONT sician. For molten maleic ykg use moderate irritation, su temporary. g of the skin and first-deg	roat irritation. Skin contact phobia and double vision rACT: flush with lots of burns, remove crust and				

	4. FIRE HAZARDS	7. SHIPPING INFORMATION
4.1	Flash Point: (Liquid) 215°F C.C.; 230°F O.C.	7.1 Grades of Purity: Commercial: 99.5% 7.2 Storage Temperature: Ambient
4.2	Flammable Limits in Air: 1.4%-7.1%	7.3 Inert Atmosphere: No requirement
.3	Fire Extinguishing Agents: Alcohol foam, dry chemical or carbon dioxide	7.4 Venting: Open
.4	Fire Extinguishing Agents Not to Be	7.5 IMO Pollution Category: D
	Used: Water or foam may cause frothing	7.6 Ship Type: 3
	Special Hazards of Combustion Products: Not pertinent	7.7 Barge Hull Type: Currently not available
4.6	Behavior in Fire: When heated above 300°F in the presence of various	8. HAZARD CLASSIFICATIONS
	materials may generate heat and carbon	8.1 49 CFR Category: Corrosive material
17	dioxide. Will explode if confined. Auto Ignition Temperature: 878°F	8.2 49 CFR Class: 8
	Electrical Hazards: Class I, Group D	8.3 49 CFR Package Group: III 8.4 Marine Pollutant: No
1.9	Burning Rate: 1.4 mm/min.	8.5 NFPA Hazard Classification:
l.10	Adiabatic Flame Temperature: Currently	
44	not available Stoichometric Air to Fuel Ratio: 14.3	Category Classification Health Hazard (Blue) 3
611	(calc.)	Flammability (Red) 1
1.12	Flame Temperature: Currently not	Instability (Yellow) 1
	available	8.6 EPA Reportable Quantity: 5000 pounds
.13	Combustion Molar Ratio (Reactant to Product): 5.0 (calc.)	8.7 EPA Pollution Category: D
1.14	Minimum Oxygen Concentration for	8.8 RCRA Waste Number: U147
	Combustion (MOCC): Not listed	8.9 EPA FWPCA List: Yes
	5. CHEMICAL REACTIVITY	9. PHYSICAL & CHEMICAL PROPERTIES
5.1	Reactivity with Water: Hot water may	
	cause frothing. Reaction with cold water is slow and non-hazardous.	9.1 Physical State at 15° C and 1 atm: Solid
5.2	Reactivity with Common Materials: No	9.2 Molecular Weight: 98.06
	reaction	<b>9.3 Boiling Point at 1 atm:</b> 392°F = 200°C = 473°K
	Stability During Transport: Stable	<b>9.4 Freezing Point:</b> 127°F = 53°C = 326°K
.4	Neutralizing Agents for Acids and Caustics: Solid spills can usually be re-	9.5 Critical Temperature: Not pertinent
	covered before any significant reaction	9.6 Critical Pressure: Not pertinent
	with water occurs. Flush area of spill with water.	9.7 Specific Gravity: 1.43 at 15°C (solid)
5.5	Polymerization: Very unlikely at ordinary	9.8 Liquid Surface Tension: Not pertinent
	temperatures, even in the molten state.	9.9 Liquid Water Interfacial Tension: Not pertinent
0.6	Inhibitor of Polymerization: None	9.10 Vapor (Gas) Specific Gravity: Not pertinent
	6. WATER POLLUTION	9.11 Ratio of Specific Heats of Vapor (Gas):
4		Not pertinent
) <b>.</b> 1 .	Aquatic Toxicity: 150 ppm/24 hr/sunfish/TLm/fresh water	9.12 Latent Heat of Vaporization: Not pertinent
6.2	Waterfowl Toxicity: Currently not available	9.13 Heat of Combustion: -5936 Btu/lb = -3298 cal/g = -138.1 X 10 <sup>5</sup> J/kg
<b>3.3</b>	Biological Oxygen Demand (BOD): 50%,	9.14 Heat of Decomposition: Not pertinent
	5 days	9.15 Heat of Solution: -153 Btu/lb = -85.0 cal/g = -3.56 X 10 <sup>5</sup> J/kg
.4	Food Chain Concentration Potential:	9.16 Heat of Polymerization: Not pertinent
6.5	GESAMP Hazard Profile:	9.17 Heat of Fusion: Currently not available
	Bioaccumulation: 0	9.18 Limiting Value: Currently not available
	Damage to living resources: 1 Human Oral bazard: 2	9.19 Reid Vapor Pressure: Currently not
	Human Contact hazard: II	available
	Reduction of amenities: XX	
	Damage to living resources: 1 Human Oral hazard: 2 Human Contact hazard: II	9.19 Reid Vapor Pressure: Currently not

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9.20 SATURATED LIQUID DENSITY		9.21 LIQUID HEAT CAPACITY		9.22 LIQUID THERMAL CONDUCTIVITY		9.23 LIQUID VISCOSITY	
Temperature (degrees F)	Pounds per cubic foot	Temperature (degrees F)	British thermal unit per pound-F	Temperature (degrees F)	British thermal unit inch per hour-square foot-F	Temperature (degrees F)	Centipoise
	N O T		N O T		N O T		N O T
	P E R T I N E N T		P E R T I N E N T		P E R T I N E N T		P E R T I N E N T

9.24 SOLUBILITY IN WATER		9.25 SATURATED VAPOR PRESSURE		9.26 SATURATED VAPOR DENSITY		9.27 IDEAL GAS HEAT CAPACITY	
Temperature (degrees F)	Pounds per 100 pounds of water	Temperature (degrees F)	Pounds per square inch	Temperature (degrees F)	Pounds per cubic foot	Temperature (degrees F)	British thermal unit per pound-F
	of water I N S O L U B L E		N O T E R T I N E N T		N O T E R T I N E N T		pound-F N O T E R T I N E N T